INVESTIGATION OF THE STRUCTURAL
PECULIARITIES AND CHEMICAL TRANSFORMATIONS OF CARBAZOLE AND ITS
DERIVATIVES

YYVI \* SYNTHESIS OF AZOMETHINES EROM

XXXI.\* SYNTHESIS OF AZOMETHINES FROM

3-AMINOCARBAZOLE

T. M. Kulikova, V. I. Shishkina, and I. P. Ryazanov

UDC 547.759.32+547.541.52

Ten new azomethine compounds, derivatives of 3-aminocarbazole and aromatic aldehydes, were synthesized. The IR spectra of the compounds were studied.

The azomethine derivatives of various classes of organic compounds have attracted attention as a result of their broad synthetic possibilities and the diversity of the paths for their practical application. The literature contains only limited data on the azomethines of the carbazole series [2].

We have synthesized a number of azomethines of 3-aminocarbazole in high yields by condensing it with equimolecular amounts of various aromatic aldehydes. The condensation was carried out in alcohol, either at room temperature or with heating. The product yield was the criterion for selection of the temperature conditions. The azomethines obtained are yellow to red substances which are satisfactorily crystallized from organic solvents. The results are presented in Table 1.

The IR spectra of the compounds were obtained in order to elucidate several structural peculiarities. An intense absorption band due to the presence of frequencies of the stretching vibrations of the NH bond for the substances in the solid state and in solutions is observed at  $3400-3450 \text{ cm}^{-1}$ ; this is probably due to the presence of an intermolecular  $=N-H \dots N=$  hydrogen bond (Table 2).

All of the azomethines have intense absorption bands at  $1612-1621 \text{ cm}^{-1}$  which are assigned to the stretching vibrations of the conjugated C = N group [4]. The absence of a band for the stretching vibrations of the OH group, even in solution, is explained by the presence of a strong intramolecular hydrogen bond with the nitrogen of the azomethine group, which is manifested in the form of a broad, diffuse band from  $2800 \text{ to } 2900 \text{ cm}^{-1}$ .

## EXPERIMENTAL

- A. An alcoholic solution of 5 g (0.027 mole) of 3-aminocarbazole in 150 ml of alcohol was mixed with equimolar amounts of the appropriate aromatic aldehyde, and the reaction mass was heated to the boiling point for 2 to 3 h. The precipitate was filtered after cooling and purified by recrystallization.
- B. An alcoholic solution of the appropriate aldehyde (0.027 mole) was added with stirring to a suspension of 5 g (0.027 mole) of 3-aminocarbazole in 50 ml of alcohol. The reaction mass was stirred for 3 to 4 h at room temperature, after which the precipitate was filtered and purified by recrystallization.

© 1973 Consultants Bureau, a division of Plenum Publishing Corporation, 227 West 17th Street, New York, N. Y. 10011. All rights reserved. This article cannot be reproduced for any purpose whatsoever without permission of the publisher. A copy of this article is available from the publisher for \$15.00.

<sup>\*</sup>For Communication XXX see [1].

G. I. Nosov Magnitogorsk Mineral-Metallurgical Institute. S. M. Kirov Ural Polytechnical Institute, Sverdlovsk. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 10, pp. 1353-1355, October, 1970. Original article submitted May 5, 1968.

TABLE 1

							ż	%	 
Comp.	Ar	Meth- od	Color	dm	Solvent	Empirical formula	punoj	calc.	Yield,
_	C <sub>e</sub> H <sub>e</sub>	4	Light-vellow	500	Acetone	CaHaN	101	10.4	 
•		 {	1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2			0.1911.1917	1	5	5
II	o-HOC <sub>6</sub> H <sub>4</sub>	¥	Yellow	223—224	n-Butanol	C <sub>19</sub> H <sub>14</sub> N <sub>2</sub> O	9,4	8'6	93
III	o-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	Р	Orange	177—180	Ethanol	C <sub>20</sub> H <sub>16</sub> N <sub>2</sub> O	0,6	6,9	7.0
N	o-HOOCC₀H₄	A	Brown	255256	Ethanol	C20H14N2O2	8,7	8,0	82
>	2-Hydroxynaphthy1	▼	Red-orange	275-276	Glacial acetic acid	$C_{23}H_{16}N_2O_2$	8,1	8,3	06
ΙΛ	2-NO <sub>2</sub> -6-HOC <sub>6</sub> H <sub>3</sub>	р	Red	> 345	Dimethylformamide	C <sub>19</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub>	12,1	12,7	88
VII	2-OH-3-NO <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	Ą	Red-orange	238239	Ethanol	C <sub>19</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub>	12,0	12,7	25
VIII	2-OH-5-NO <sub>2</sub> C <sub>6</sub> H <sub>8</sub>	4	Orange	283—285	2	C <sub>19</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub>	12,3	12,7	87
XI	2-OH-5-BrC6H3	V	Yellow	278—281	\$	C <sub>19</sub> H <sub>13</sub> N <sub>2</sub> OBr	7,3	7.7	94
×	9-Anthraceny1	A	Yellow	218—220	Acetone	$C_{27}H_{19}N_2$	7,2	2,6	72

TABLE 2. Frequencies of the Stretching Vibrations of the NH Bond for Several Azomethines

	Frequency v, cm <sup>-1</sup>		
Compound	in mineral oil	in dichloro- ethane	Δv. cm <sup>-1</sup>
I II V VIII	3410 3400 3405 3409	3448 3450 3450 3445	38 50 45 36

The IR spectra of the azomethines in the solid state (mineral oil paste) and in dichloroethane solution were obtained with a UR-10 spectrophotometer.

## LITERATURE CITED

- 1. T. I. Proshechkina, V. I. Shishkina, N. D. Negodyaev, and G. M. Novikova, Izv. Vuzov., Ser. Khim., 12, 903 (1969).
- 2. Ng. Buu-Hoi and Ng. Hoah, J. Org. Chem., 16, 1327 (1951).
- 3. L. Bellamy, Infrared Spectra of Complex Molecules, Methuen (1958).
- 4. N. Jones and C. Sandorfy, Application of Spectroscopy to Chemistry [Russian translation], IL, Moscow (1959), p. 209.